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web correct answer option b q3 the image shows some chemical reactions which option identifies the reactants and products of the reactions correct answer option c q4 a student performs an experiment to form aluminium chloride from aluminium and chlorine

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web apr 19 2023 chemical reactions and equations extra questions very short answer type question 1 why is the combustion of liquified petroleum gas lpg a chemical change answer combustion of liquified petroleum gas lpg is a chemical change because after its combustion a new substance is formed and cannot be turned back into

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web q5 identify the reducing agent in the following reactions a $4\text{nh}_3 + 5\text{o}_2 \rightarrow 4\text{no} + 6\text{h}_2\text{o}$ b $\text{h}_2 + \text{f}_2 \rightarrow 2\text{hf}$ c $\text{fe}_2\text{o}_3 + 3\text{co} \rightarrow 2\text{fe} + 3\text{co}_2$ d $2\text{h}_2 + \text{o}_2 \rightarrow 2\text{h}_2\text{o}$ answer a here ammonia nh_3 is the reducing agent b here water h_2o is the reducing agent