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seaborgium determine if the following electron configurations are correct 26 1s22s22p63s23p64s24d104p65s1 no it should be 3d10

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each isotope answer co has 27 protons 27 electrons and 33 neutrons 1s 2 2s 2 2p 6 3s 2 3p 6 4s 2 3d 7 i has 53 protons 53

electrons and 78 neutrons 1s 2 2s 2 2p 6 3s 2 3p 6 3d 10 4s 2 4p 6 4d 10 5s 2 5p 5

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web select the electron configuration tab on show number of electrons introduction beyond argon it is a bit tricky to

determine which subshell gets filled next there are several rules that scientists use to determine the electron configurations

of larger atoms question how are the electron configurations of elements beyond argon determined