

# SOME THERMODYNAMIC RELATIONS

## HEAT TRANSFER

Conduction  $Q = kA dt/dx$   $k$  is the thermal conductivity

Convection

Radiation  $P = \epsilon\sigma T^4$  per unit area.  $\epsilon$  is the emissivity,  
 $\sigma$  is the Stefan-Boltzmann constant  
 $= 5.67 \cdot 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$

1<sup>st</sup> Law  $\Delta U = \bar{d}Q - \bar{d}W$  For fluids  $\bar{d}W = pdV$

Entropy  $\Delta S = \Delta Q_{REV} / T$

Efficiency of a Carnot Engine  $\eta_C = 1 - T_2 / T_1$

Enthalpy  $h = u + pv$  Useful for describing

1. Heat transfer at constant pressure in a boiler or  
condenser. Heat input  $Q = h_2 - h_1$  (2.1)

2. Adiabatic changes in compressors and turbines where  
net work on the shaft  $W = h_1 - h_2$  (2.2)

(See derivation 2.2 in the text book for proof of these last two relations)

# Thermodynamic Relations In Multi Compone

**Thomas Griffiths**



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