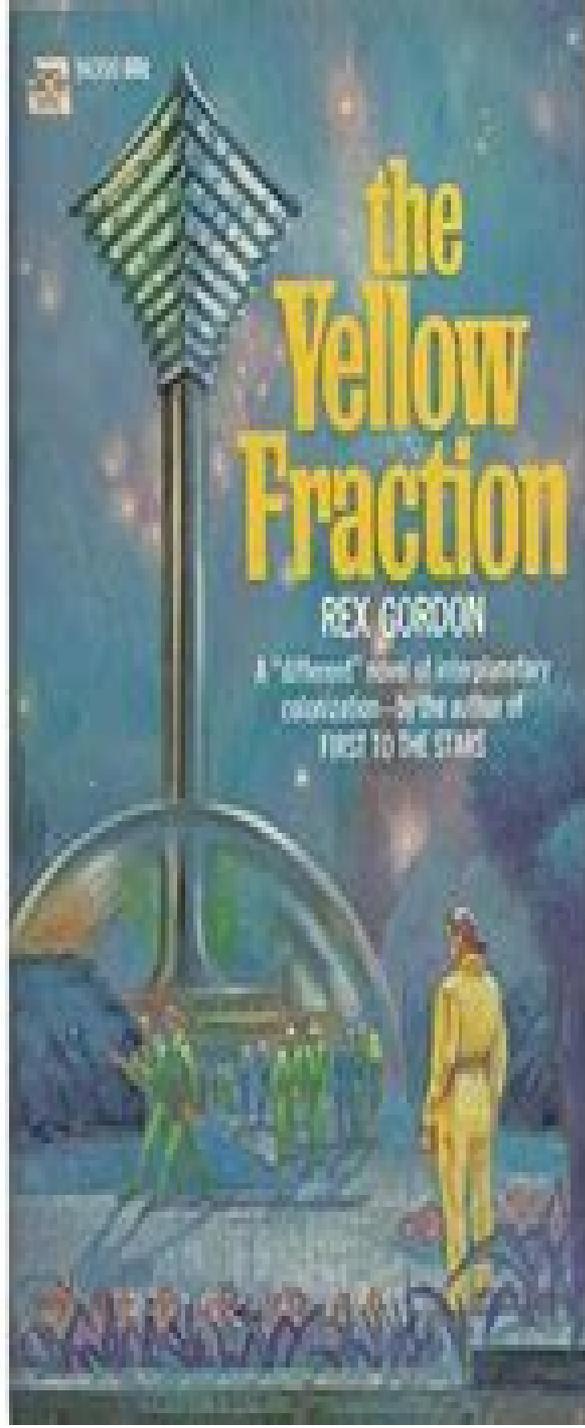


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Electrochemical determination of the Gibbs energy of formation of sphalerite (ZnS) Seth C. Schaefer, 1978 **Report of the Board, with the Report of the Director of Food Investigation** Great Britain. Food Investigation Board, 1929 Colorado School of Mines Quarterly, 1924 **Quarterly of the Colorado School of Mines ...** Colorado School of Mines, 1921 **Cobalt and Nickel Removal from Zinc Sulfate Electrolyte by Solvent Extraction and Precipitation Techniques** Richard G. Sandberg, Terry L. Hebble, 1978 Sphalerite concentrates prepared during processing of Missouri lead ores contain small percentages of cobalt and nickel which adversely affect zinc electrolysis The Bureau of Mines has evaluated solvent extraction and precipitation techniques to remove and recover cobalt and nickel from zinc sulfate solution prior to zinc electrolysis Prepared zinc sulfate solutions containing about 190 gpl of zinc and 50 ppm of both cobalt and nickel were treated using various combinations of complexing reagents solvent pH concentration time and temperature The following complex reagents gave the best results a nitroso naphthol nitrosos a naphthol 1 2 cyclohexane dione dioxime nioxime di 2 pyridyl ketone oxime and dimethglyoxime These reagents in suitable solvents lowered the cobalt and nickel levels to **Bibliotics** Persifer Frazer, 1901 **American Journal of Pharmacy and the Sciences Supporting Public Health**, 1916 *American Journal of Pharmacy*, 1916 **Bulletin of the Chemical Society of Japan** Nihon Kagakkai, 1928 The Pharmaceutical Era, 1888 *Journal of the Chemical Society* Chemical Society (Great Britain), 1926

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