

The Devil In God's Old Man

by Sue Mayes Newhall

Life and work with the Burma Surgeon,
Gordon S. Seagrave



The Devil In Gods Old Man

Jiyuan Zhang



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with 1200 ml of 2M HNO₃ if the final density of solution is same as pure water then molarity of the iron in resulting solution which decides the nature of the above solution is 1) 0.25M 2) 0.5M 3) 0.888M 4) none of

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$$m = \frac{M \cdot 1000}{1000 - d \cdot M}$$

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solution of molarity 1.5 M requires 1.5 mol of Na to every litre of solvent

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$$M_{mix} = \frac{m_1 \cdot V_1 + m_2 \cdot V_2 + \dots + m_n \cdot V_n}{V_{total}}$$

where m is mass and V is volume. For example, if you have 480 g of NaCl in 2 L of solution, the molarity is $\frac{480}{2} = 240$ g/L. With an increase in temperature, the volume of the solution increases, which decreases the molarity.

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