



VOICE
OF
CONSCIENCE

Voice Of Conscience

Chao Zhang



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web write the oxidation and reduction reactions for each redox reaction the first one is done for you $23 \text{ Fe} + \text{Co} \rightarrow \text{Co} + 2 \text{ Fe}$
oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2e^{-}$ reduction $\text{Fe}^{2+} + 2e^{-} \rightarrow \text{Fe}$ $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni} + 3 \text{ Ag}$ oxidation reduction $25 \text{ Cu} + \text{Pb} \rightarrow \text{Pb} + \text{Cu}$ oxidation reduction 26 O_2

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent $\text{a} \text{ I}_2 + \text{O}_2 \rightarrow \text{I}_2\text{O}_5$
 $5 \text{ CO} + \text{I}_2 \rightarrow 5 \text{ CO}_2 + \text{I}_2$ $5 \text{ C} + \text{O}_2 \rightarrow 5 \text{ CO}_2$

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