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## STARTING WITH SAFETY

*Use this worksheet to take notes while watching the ACS Safety. You can refer to this worksheet to review safety information throughout the semester. You may watch the video as many times as you need.*

### Handling Chemicals Safety

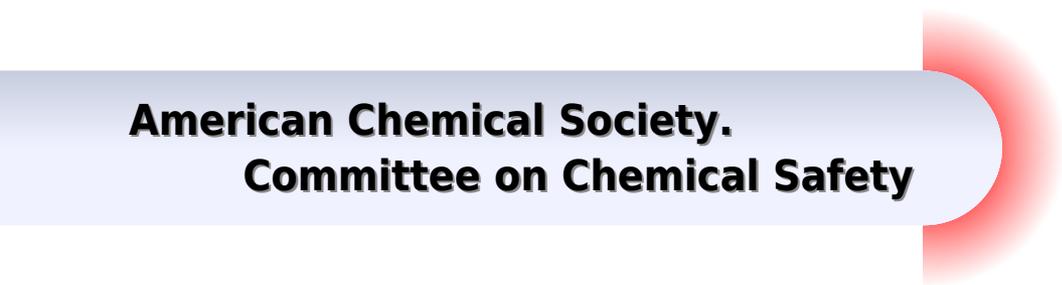
1. What size containers should be used for pouring? Small containers that you can control easily
2. How many times should you read the label information before you open the container?  
Read and reread
3. Is it important to be exact when following the instructions? yes
4. When diluting acid, do you add acid to water or add water to acid? Add the acid to the water
5. Where should you place your hand when pouring from bottles? Hold the bottles with your hand over the label
6. If you are given permission, what is the proper way to smell a chemical in the lab? Hold it away from your face and waft the vapors
7. Should you use glassware if it is dirty? No
8. What should you do with leftover chemicals? Dispose of it in the proper waste container
9. How can you safely draw small volumes of liquids into a pipette?  
Don't use your mouth use a pipettor
10. If you spill a liquid chemical, what should you do? Report the spill immediately, stay away from the spilled chemical and let the teacher clean it up
11. What three things should be cleaned up at the end of each lab? Glassware should be scrubbed and rinsed, liquid puddles or powders left behind should be wiped with a paper towel and placed in the proper container, clean any chemicals off your hands before leaving

### Bunsen Burner and Glassware Safety

1. What does the valve on the bottom of the Bunsen burner control? The flow of gas
2. Before hooking a Bunsen burner to the gas line, what should you look for? Check the gas hose for cracks and make sure the hose fits securely on the gas valve and on the Bunsen burner
3. What two things can be used to light the burner? Striker and match
4. What is the hottest part of the flame? The tip of the inner cone
5. If your flame sputters or goes out, what should you do? Turn the gas off immediately
6. If you smell gas, what should you do? Turn it off
7. Before you heat glassware, what should you check first? check for cracks or stars

# Starting With Safety Introduction For The Academic Chem Labplus Video

**American Chemical Society.  
Committee on Chemical Safety**



## **Starting With Safety Introduction For The Academic Chem Labplus Video:**

**Introduction to Safety in the Chemical Laboratory** N. T. Freeman, J. Whitehead, 1982      **Introduction to Chemical Laboratory Safety** George D. Heindel, American Chemical Society, 1991      *Laboratory Safety for Chemistry Students* Robert H. Hill, Jr., David C. Finster, 2016-04-21 Provides knowledge and models of good practice needed by students to work safely in the laboratory as they progress through four years of undergraduate laboratory work Aligns with the revised safety instruction requirements from the ACS Committee on Professional Training 2015 Guidelines and Evaluation Procedures for Bachelor's Degree Programs Provides a systematic approach to incorporating safety and health into the chemistry curriculum Topics are divided into layers of progressively more advanced and appropriate safety issues so that some topics are covered 2-3 times at increasing levels of depth Develops a strong safety ethic by continuous reinforcement of safety to recognize, assess and manage laboratory hazards and to plan for response to laboratory emergencies Covers a thorough exposure to chemical health and safety so that students will have the proper education and training when they enter the workforce or graduate school      [Safety in academic chemistry laboratories](#), 2003 This book contains volume 1 of 2 and describes safety guidelines for academic chemistry laboratories to prevent accidents for college and university students Contents include 1 Your Responsibility for Accident Prevention 2 Guide to Chemical Hazards 3 Recommended Laboratory Techniques and 4 Safety Equipment and Emergency Procedures Appendices include the Web as a source of safety information and incompatible chemicals      **Starting with Safety (\*vhs/ Pal Videotape\*) Cat #v-5100** Lois Wickstrom, American Chemical Society, Elizabeth J. Vaughn, Robert Kaper, Andy Hemmendinger, Bob Burnett, Janet Shalestik, Andy Joyce, American Chemical Society. Continuing Education Department, 1991 Provides an overview on handling chemicals and equipment safely proper lab behavior and safety techniques      *Safety in Academic Chemistry Laboratories* American Chemical Society. Committee on Chemical Safety, 1979      **Safety in Academic Chemistry Laboratories** American Chemical Society, 1979      **Safety in the Chemical Laboratory and Industry** Muhammad Raza Shah, 2023-07-11 Safety in the Chemical Laboratory and Industry A Practical Guide provides an authoritative reference on chemical safety procedures and the handling and disposal of chemicals at both the laboratory and industrial level The book serves as a practical guide for laboratory workers and offers prudent guidelines for the development of regulatory policy for academic and non-academic organizations The book stresses that all accidental injuries can be minimized if systems and attitudes are in place to prevent them and that to fully materialize a safety culture even with improved technology everyone should be mindful of maintaining a safe environment This handbook will assist faculty, researchers, staff and students to provide a safe and healthy laboratory environment to teach, learn and conduct research Helps understand the basic principles of chemical safety Provides guidance how to create a culture of chemical safety and how to deal with hazardous waste disposal Handles minimization of risks related to chemicals      **Safety in Academic Chemistry Laboratories** American Chemical Society. Committee on Chemical Safety, 2017      *School*

*Chemistry Laboratory Safety Guide*, 2006      **School Chemistry Laboratory Safety Guide** Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute Health and Safety, 2014-02-16 In 1984 the Council of State Science Supervisors in association with the U S Consumer Product Safety Commission and the National Institute for Occupational Safety and Health published the safety guide *School Science Laboratories A Guide to Some Hazardous Substances* to help science teachers identify hazardous substances that may be used in school laboratories and provide an inventory of these substances Because school science curricula have changed since then the safety guide has been updated and revised to reflect those changes This guide on safety in the chemistry laboratory was also written to provide high school chemistry teachers with an easy to read reference to create a safe learning environment in the laboratory for their students The document attempts to provide teachers and ultimately their students with information so that they can take the appropriate precautionary actions in order to prevent or minimize hazards harmful exposures and injuries in the laboratory The guide presents information about ordering using storing and maintaining chemicals in the high school laboratory The guide also provides information about chemical waste safety and emergency equipment assessing chemical hazards common safety symbols and signs and fundamental resources relating to chemical safety such as Material Safety Data Sheets and Chemical Hygiene Plans to help create a safe environment for learning In addition checklists are provided for both teachers and students that highlight important information for working in the laboratory and identify hazards and safe work procedures This guide is not intended to address all safety issues but rather to provide basic information about important components of safety in the chemistry laboratory and to serve as a resource to locate further information

**CHEMICAL SAFETY IN THE LABORATORY.** STEPHEN K. HALL, 2023      **Improving Safety in the Chemical Laboratory** Jay A. Young, 1991-06-26 The work of accident prevention in the lab begins with foresight Discerning close calls near accidents early enough prevents them from turning into full fledged mishaps mishaps that cost time and money and which could result in injury Improving Safety in the Chemical Laboratory is an accident prevention handbook for the professional in the lab that shows how to detect and eliminate the causes of dangerous mishaps and virtually hazard proof any lab environment In unequivocally clear and practical terms Improving Safety in the Chemical Laboratory Second Edition offers detailed procedures from precautionary labeling to simulated drills safety inspections and the preparation of a chemical hygiene plan for the development of a safety enhanced workplace Reflecting in part the upgraded procedures now mandated by the OSHA Laboratory Standard in the USA as well as the WHMIS regulations in Canada and the COSHH regulations in the United Kingdom this newest edition offers unparalleled and up to date guidance on the fine points of hazard control with new added material on managing and handling especially hazardous substances and personal protective equipment The 95 percent solution the list of causes of laboratory accidents Hazard categories unsafe acts unsafe conditions Selecting and maintaining personal protective conditions Accident handling Classes of fuels and fires Preventing and

extinguishing fires Toxic effects of chemicals Recognition of and treatment for exposure Chemical specific safety protocol Storage of lab chemicals Safe disposal of hazardous waste Personal protective equipment in the laboratory Improving hood performance Designing safety into new or renovated laboratories A comprehensive one volume safety seminar Improving Safety in the Chemical Laboratory will provide indispensable guidance to lab supervisors and workers teachers and students and anyone involved in the investigation of chemical accidents and injury In clear language that quickly details the full range of hidden and avoidable laboratory hazards Improving Safety in the Chemical Laboratory Second Edition offers the most up to date practical and easy to implement lab safety regimen yet available *Safety in Academic Chemistry Laboratories*, 1990 *Guidelines for Chemical Laboratory Safety in Academic Institutions* American Chemical Society. Committee on Chemical Safety, 2016 To assist educators with integrating safety education throughout the entire chemistry curriculum where principles of safety are taught over time rather than just during one time safety trainings ACS has published new Guidelines for Chemical Laboratory Safety in Academic Institutions The Safety Education Guidelines are organized around the concept of R A M P an acronym for the Four Principles of Safety Recognize the hazard Assess the risk of the hazard Minimize the risk of the hazard and Prepare for emergencies The guidelines also include student learning outcomes statements which clearly state the expected knowledge skills attitudes competencies in the area of chemical safety that students are expected to acquire as they progress with their education *Laboratory Safety Manual* University of Bristol. School of Chemistry, 2000 **School Chemistry Laboratory Safety Guide** Department of Health and Human Services, 2013-06-22 Recognition of laboratory safety and health problems has crystallized since the passage of the Occupational Safety and Health Act of 1970 This Act requires that certain precautions be observed to protect the safety and health of employees on the job The employee designation includes all teachers employed by private and public school systems in States that have occupational safety and health plans accepted by the Occupational Safety and Health Administration OSHA of the U S Department of Labor DOL OSHA rules and regulations are provided to protect the employees and the facilities The importance of laboratory safety has been recognized for many years in industry However educational institutions have been slower to adopt such safety practices and programs **Safety Manual for the Academic Chemistry Laboratory** Adelaide Dittmar, 1983

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the fertilization brings together 23 chromosomes from the male and 23 chromosomes from the female resulting in the formation of a fertilized egg cell with 46 chromosomes the fertilized

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choose 1 answer the egg cell is the ovum the sperm cell fuses with it directly a the egg cell is the ovum the sperm cell fuses with it directly the oocyte inside the egg cell divides to

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fertilisation happens when an egg cell meets with a sperm cell and joins with it the fertilised egg divides to form a ball of cells called an embryo the embryo attaches to the lining of the

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offspring that are genetically unique fertilisation is the process in which the nucleus of a sperm cell fuses with the nucleus of an egg cell to produce a zygote which will eventually grow into

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feb 12 2018 instructor what we re gonna do with this video is talk about fertilization and development in human beings or at least early development in human beings and this right

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key points a multicellular organism develops from a single cell the zygote into a collection of many different cell types organized into tissues and organs development involves cell

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jul 30 2022 describe the obstacles that sperm must overcome to reach an oocyte explain capacitation and its importance in fertilization summarize the events that occur as a sperm

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the early stages of embryonic development are also crucial for ensuring the fitness of the organism fertilization fertilization pictured in figure 43 23a is the process in which

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fertilisation occurs when a sperm and egg join to form an embryo an embryo develops into an unborn baby in the uterus during pregnancy

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