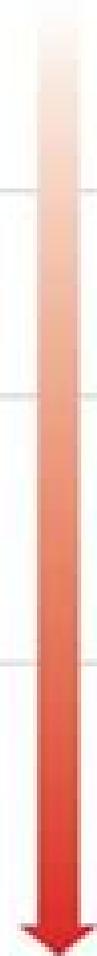


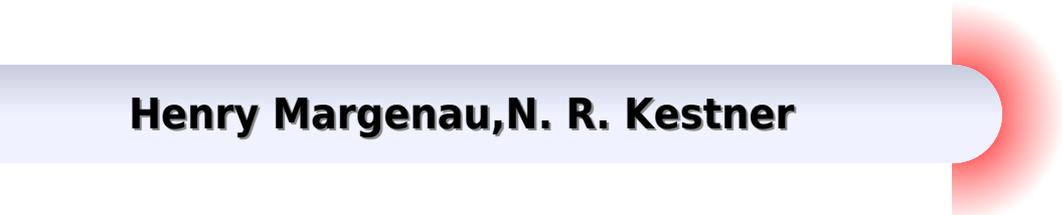
# Intermolecular forces

TABLE 11.4 Types of Intermolecular Forces

Type	Present in	Molecular perspective	Strength
Dispersion	All molecules and atoms		
Dipole-dipole	Polar molecules		
Hydrogen bonding	Molecules containing H bonded to F, O, or N		
Ion-dipole	Mixtures of ionic compounds and polar compounds		

# Theory Of Intermolecular Forces

**Henry Margenau, N. R. Kestner**



## Theory Of Intermolecular Forces:

**Theory of Intermolecular Forces** H. Margenau, N. R. Kestner, 2013-10-22 Theory of Intermolecular Forces deals with the exposition of the principles and techniques of the theory of intermolecular forces The text focuses on the basic theory and surveys other aspects with particular attention to relevant experiments The initial chapters introduce the reader to the history of intermolecular forces Succeeding chapters present topics on short intermediate and long range atomic interactions properties of Coulomb interactions shape dependent forces between molecules and physical adsorption The book will be of good use to experts and students of quantum mechanics and advanced physical chemistry

**Theory of Intermolecular Forces** Henry Margenau, N. R. Kestner, 1969 **The Theory of Intermolecular Forces** Anthony J. Stone, 2013-01-31 The Theory of Intermolecular Forces sets out the mathematical techniques needed to describe and calculate intermolecular interactions in physics and chemistry and to handle the more elaborate mathematical models used to represent them *The Theory of Intermolecular Forces* Anthony J. Stone, 1996 Describes advances in the theory of intermolecular forces and sets out the mathematical techniques that are needed to handle the more elaborate models that are being used increasingly by both theoreticians and experimentalists Includes a detailed account of the use of higher rank multipole moments to describe electrostatic interactions including treatment of both Cartesian and spherical tensor methods Modern ab initio perturbation theories of intermolecular interactions are also described Annotation copyright by Book News Inc Portland OR

*Perturbation Theory of Intermolecular Forces Using an "effective" Hamiltonian* A. Ažman, B. Lukman, A. Ocvirk, 1969

Theory of Intermolecular Forces - Revised and Enlarged Margenau H., 1971 Intermolecular Forces and Their Evaluation by Perturbation Theory P. Arrighini, 2012-12-06 The aim of these notes is to offer a modern picture of the perturbative approach to the calculation of intermolecular forces The point of view taken is that a perturbative series truncated at a low order can provide a valuable way for valuating interaction energies especially if one limits oneself to the case of intermediate and long range distances between the interacting partners Although the situation corresponding to short distances is essentially left out from our presentation the problems which are within the range of the theory form a vast and important class a large variety of phenomena of matter in fact depends on the existence of interactions among atoms or molecules which over a substantial range of distances should be classified as weak in comparison to the interactions occurring inside atoms or molecules We are aware of the omission of some topics which in principle could have been included in our review For instance a very scarce attention has been paid to the analysis of problems involving interacting partners in degenerate states which is of particular relevance in the case of interactions between excited atoms only a rather quick presentation of the formal apparatus of degenerate perturbation theory is included in Chap III Interactions involving the simultaneous presence of more than two atoms or molecules have not been considered with the consequent non-necessity of considering nonadditive effects which characterize the general N body problem **Bulletin of**

**Thermodynamics and Thermochemistry**, 1977     **A Treatise on Electrical Theory and the Problem of the Universe**  
George William von Tunzelmann, 1910     **Intermolecular Forces** Geoffrey C. Maitland, 1981     *Colloid Chemistry,*  
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Stephen G. Brush, Rose Kraft, Jean Senkin, 1963     **The Chemical News and Journal of Industrial Science**, 1917  
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