

Plutonium

atomic number — 94

atomic weight — (244)

symbol — Pu

acid-base properties of higher-valence oxides —

electron configuration — [Rn]5f⁶7s²

physical state at 20 °C (68 °F) —

name — plutonium

Actinide elements

Synthetically prepared

Equal relative strength

() indicates the mass of the longest-lived isotope.

The Plutonium Standard

**American Society for Testing and
Materials**



The Plutonium Standard:

The Plutonium Standard Charles Turek, 2004-06-01 In a perfect world technology would rise to the needs of a free society and provide the means for sustenance and defense of that society A contemporary example is atomic energy which provided both the means to win a world war and a supply of cheap energy for the luxuries of growing western economies Should the discredited promise of nuclear power be called upon to throw off the fears of proliferation and tackle the problem of short term energy shortages and high fuel prices Can we trust ourselves in a free society to manage this awesome responsibility The Plutonium Standard is a cautionary tale of how one man s dream of technological superiority could turn the promise of cheap and unlimited fuel supplies into a trading game leading to ultimate destruction

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Conference on Measurement Technology for Safeguards and Materials Control, November 27-30, 1979, Kiawah Island, Charleston, South Carolina, 1979 **Compilation of American Nuclear Society Standards, 1977** American Nuclear Society, 1977 *Toxicological Profile for Plutonium*, 1990 **Measurements and Standards of Radioactivity** National Research Council (U.S.). Committee on Nuclear Science, 1958 *Analytical Chemistry in Nuclear Reactor Technology* Conference on Analytical Chemistry in Nuclear Reactor Technology, **Annual Conference on Mass Spectrometry and Allied Topics**, 1986

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