

THESES SABBATICAE, OR, THE DOCTRINE OF THE SABBATH

Thomas Shepard

Wherein the Sabbaths I. Morality, II. Change, III. Beginning, IV.
Sanctification, Are Clearly Discussed: Which Were First
Handled More Largely in Sundry Sermons in Cambridge in
New-England. ...

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calculated mass of the product expected $\times 100$ in this reaction a yield 80 is anticipated

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g Al $\times 1$ mol Al $\times 2$ mol $\text{Al}_2\text{O}_3 \times 101.964$ g $\text{Al}_2\text{O}_3 = 302.3$ g Al_2O_3 26.982 g Al 4 mol Al 1 mol Al_2O_3 c determine the percent yield
yield actual $\times 100 = 260.0$ g $\times 100 = 86.01$ theoretical 302.3 g

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C is the limiting reagent 4.33 g of H 2 are left over 13

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 CH_3OH can be produced by the following reaction $2\text{H}_2 + \text{CO} \rightarrow \text{CH}_3\text{OH}$ calculate the theoretical yield of CH_3OH if 68.5 g of CO is
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