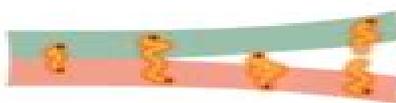


# Functional Binders



## 1. High mass loading

- **Adhesion** → Stronger interfacial interaction (e.g. hydrogen bond, Coulombic attraction,  $\pi$ - $\pi$  stacking)
- **Strength** → Physical or chemical crosslink (e.g. double helix, or organic and inorganic crosslinker))



Layer detachment



Layer cohesion

## 2. Drastic volume change

- **Disperse stress** → Sacrificial bonds & dendrimers
- **Buffer stress** → High elasticity and toughness
- **Overcome repeated expansion and contraction** → Self-healing ability from reversible interaction



Electrode delamination



Particle pulverization

## 3. High voltage

- **Inhibit migration of TM ions** → Polar and electron-rich functional groups (e.g.  $-\text{CN}$ )
- **Scavenge free radical and reactive oxygen** → Piperidine-based polymer



Electrode-electrolyte layers

## 4. Large electric impedance

- **Conjugate polymeric backbone** → Polyfluorene, polythiophene, polyaniline, polypyrrole
- **Graft conductive fragment** → Pyrene

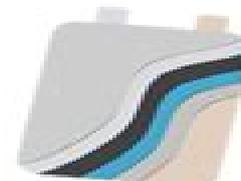
## 5. Large ionic impedance

- **Decrease crystallinity** → Increase graft degree
- **Graft flexible or ionized fragments** → Ethylene oxide or lithium sulfonate and lithium carboxylate



## 6. Thermal safety

- **Enhance thermal stability** → Inorganic binder & polyimide-based binder
- **Flame-retardant binder** → Rich in N and P elements (e.g. phosphazene)



## 7. Compatibility in special process

- **Aqueous binder** → Polar and easy to form hydrogen bonds (e.g.  $-\text{OH}$ ,  $-\text{COOH}$ ,  $-\text{NH}_2$ )
- **Weak-polarity binder** → Alkane and arene
- **Binder for solvent-free electrodes** → Thermal dispersibility



Hot Pressing

Solvent-free electrodes

# The High Binders

**Sabine Zange**



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## **Table of Contents The High Binders**

1. Understanding the eBook The High Binders
  - The Rise of Digital Reading The High Binders
  - Advantages of eBooks Over Traditional Books
2. Identifying The High Binders
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an The High Binders
  - User-Friendly Interface
4. Exploring eBook Recommendations from The High Binders
  - Personalized Recommendations
  - The High Binders User Reviews and Ratings
  - The High Binders and Bestseller Lists
5. Accessing The High Binders Free and Paid eBooks
  - The High Binders Public Domain eBooks
  - The High Binders eBook Subscription Services
  - The High Binders Budget-Friendly Options
6. Navigating The High Binders eBook Formats

- ePub, PDF, MOBI, and More
  - The High Binders Compatibility with Devices
  - The High Binders Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of The High Binders
    - Highlighting and Note-Taking The High Binders
    - Interactive Elements The High Binders
  8. Staying Engaged with The High Binders
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers The High Binders
  9. Balancing eBooks and Physical Books The High Binders
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection The High Binders
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine The High Binders
    - Setting Reading Goals The High Binders
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of The High Binders
    - Fact-Checking eBook Content of The High Binders
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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