



Sleep Optimization Manual

Rosina Ehmann



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what do the following terms mean oxidation oxidation is a loss of electrons reduction reduction is a gain of electrons what is a helpful way to remember this

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you $23 \text{ Fe} + \text{Co} \rightarrow \text{Co} + 2 \text{Fe}$
oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2e^-$ reduction $\text{Fe}^{2+} + 2e^- \rightarrow \text{Fe}$ $24 \text{ Ag} + \text{Ni} \rightarrow \text{Ni} + 3 \text{ Ag}$ oxidation reduction $25 \text{ Cu} + \text{Pb} \rightarrow \text{Pb} + 2 \text{ Cu}$ oxidation reduction 26 O_2

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