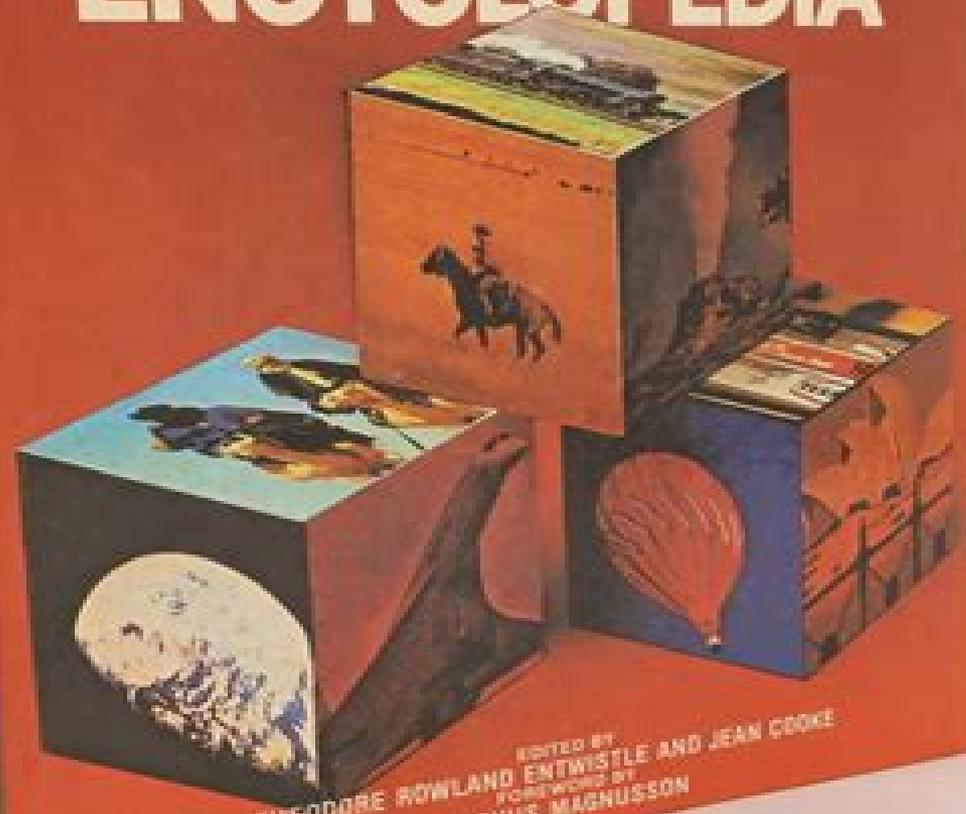


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**KJ Lindholm-Leary**



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web balance the following equations and indicate the type of reaction taking place 1  $\text{NaBr}$   $\text{H}_3\text{PO}_4$   $\text{HBr}$   $\text{Na}_3\text{PO}_4$  type of reaction 2  $\text{CaOH}$  2  $\text{Al}_2\text{SO}_4$  3  $\text{CaSO}_4$   $\text{AlOH}$  3 type of reaction 3  $\text{MgFe}_2\text{O}_3$   $\text{Fe}$   $\text{MgO}$  type of reaction 4  $\text{C}_2\text{H}_4$   $\text{O}_2$   $\text{CO}_2$   $\text{H}_2\text{O}$  type of reaction 5  $\text{PbSO}_4$   $\text{PbSO}_3$   $\text{O}_2$  type of reaction 6  $\text{NH}_3$  12