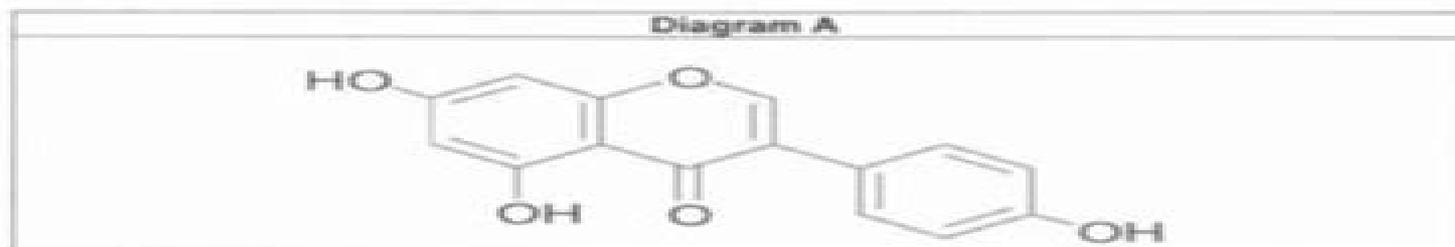


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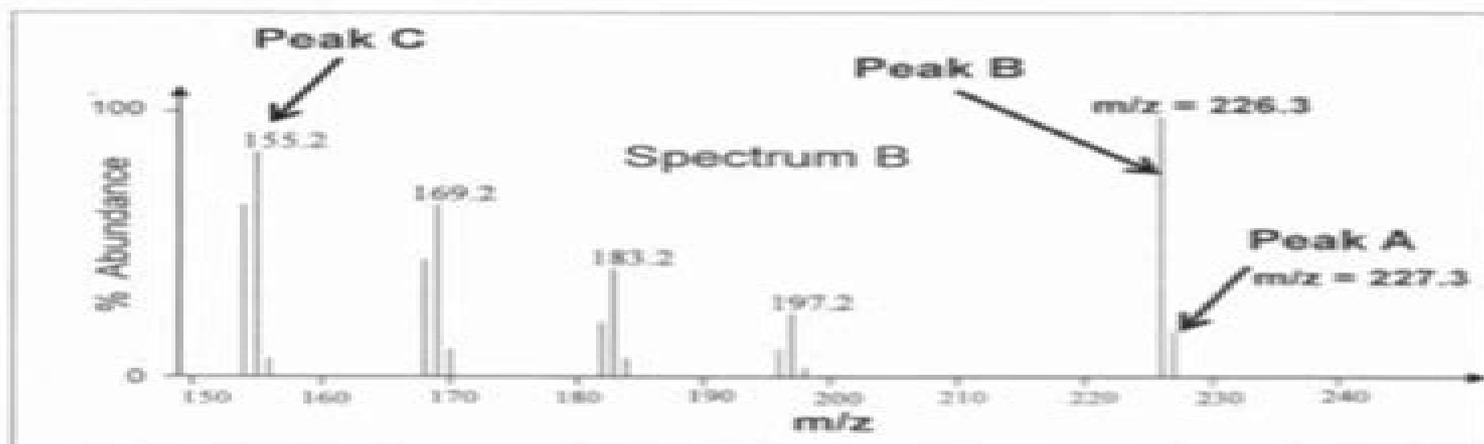
- a) Two types of mass spectrometer instruments are available, one is equipped with EI and another with ESI. You are required to analyze a compound in **Diagram A**.



Discuss the types of ionization suitable for the molecular weight determination and fragmentation information for the above compound.

(4 marks)

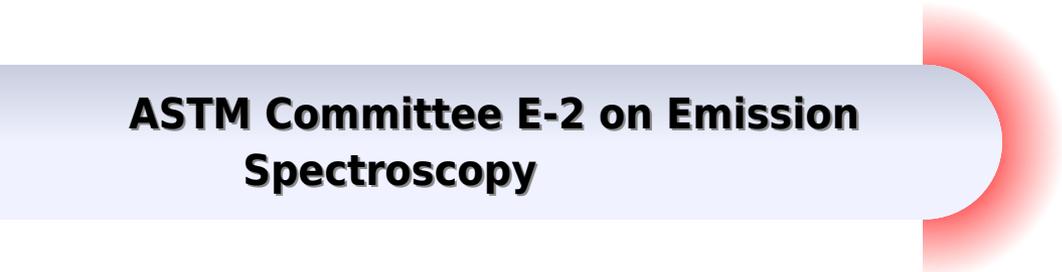
- b) Mass spectrum of pure $n\text{-C}_{10}\text{H}_{22}$ below (Spectrum B) is acquired in the range of $m/z=150\text{-}250$ using one of the ionization techniques in mass spectrometry. Choose the base peak from the mass spectrum and calculate the relative abundance of peak A. [Relative abundances: $^1\text{H} = 100\%$, $^2\text{H} = 0.016\%$, $^{12}\text{C} = 100\%$, $^{13}\text{C} = 1.08\%$]



(3 marks)

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Methods for Emission Spectrochemical Analysis ASTM Committee E-2 on Emission Spectroscopy,1968 **A.S.T.M.**
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Proceedings of the Seminar on Geochemical Prospecting Methods and Techniques United Nations. Economic Commission for Asia and the Far East,1963 **Chemical Engineering and Mining Review** ,1950 *Future Trends in Spectroscopy* ,1990

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