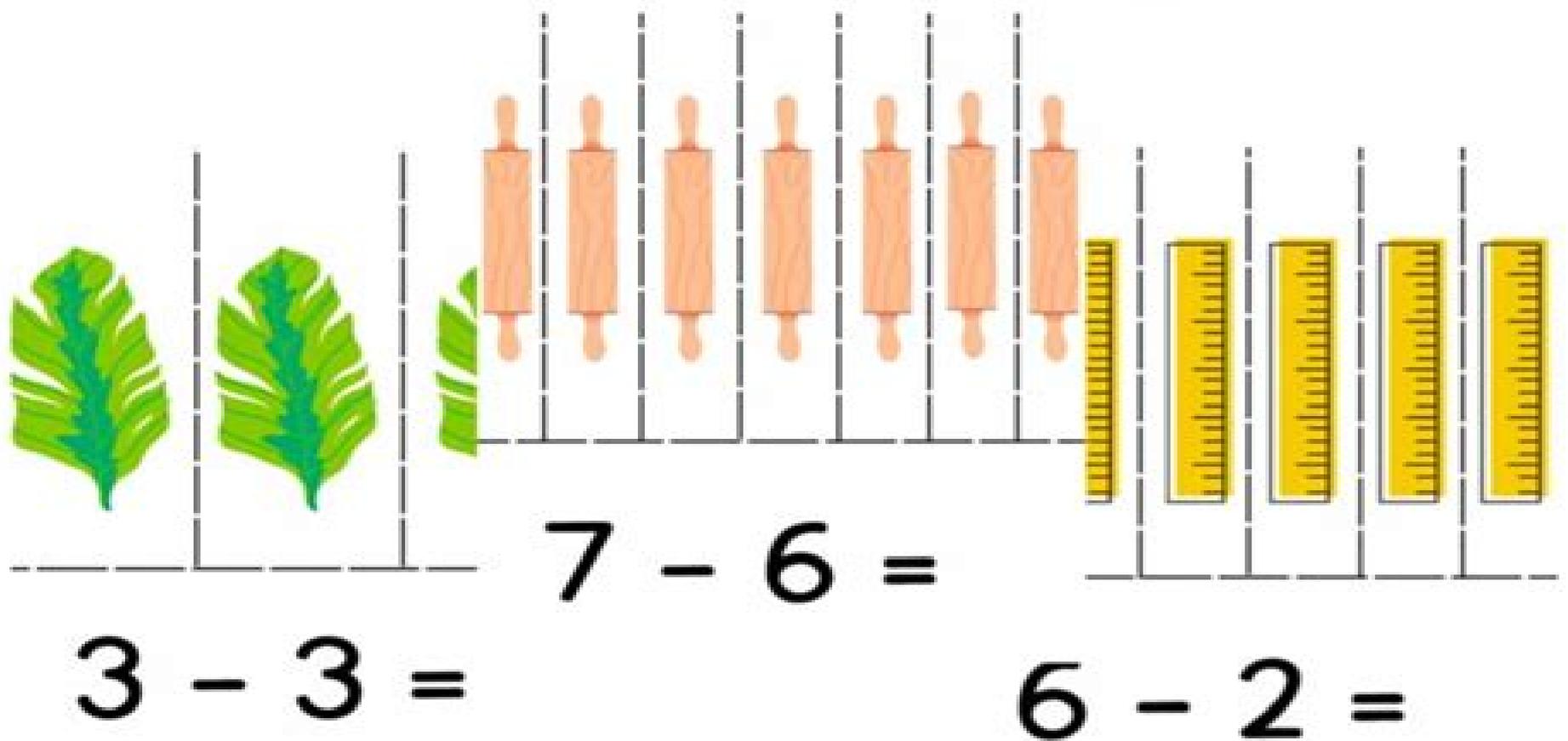


# Subtraction Action



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**Shasha Hu**



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6 50 g Al  $\times$   $\frac{2 \text{ mol Al}_2\text{O}_3}{4 \text{ mol Al}}$   $\times$   $\frac{101.96 \text{ g Al}_2\text{O}_3}{1 \text{ mol Al}_2\text{O}_3}$  = 4 mol Al  $\times$  2 mol

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