

Weber's electrodynamics for the hydrogen atom

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ABSTRACT

The original Weber action at a distance theory is valid for slowly varying effects, and it in addition to predicting all of the usual electro-dynamical results, leads to crucial effects where the Maxwell theory fails. The Weber's approach is an alternative to Maxwell electrodynamics, where the Coulomb's law becomes velocity dependent [1-6]. Here we prove that the Weber's theory gives the fine structure energy level splitting for the hydrogen atom without the assumption of mass change with velocity.

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INTRODUCTION

In Maxwell electrodynamics, Newton's third law does not hold for particles. Instead, particles exert forces on electromagnetic fields, and fields exert forces on particles, but particles do not *directly* exert forces on other particles. Therefore, two nearby particles need not experience equal and opposite forces. Related to this, Maxwell electrodynamics predicts that the laws of conservation of momentum and conservation of angular momentum are valid *only* if the momentum of particles *and* the momentum of surrounding electromagnetic fields are taken into account. The total momentum of all particles is not necessarily conserved, because the particles may transfer some of their momentum to electromagnetic fields or vice-versa. The well-known phenomenon of radiation pressure proves that electromagnetic waves are indeed able to "push" on matter, see Maxwell stress tensor and Poynting [7] vector for further details. The Weber force law is quite different: All particles, regardless of size and mass, will exactly follow Newton's third law. Therefore, Weber electrodynamics, unlike Maxwell theory, has conservation of *particle* momentum and conservation of *particle* angular momentum.

The main point to be noted about the Lagrangian method is that when potential energy depends only on position of the particle described (here we shall simplify by considering a one-particle system) there is a direct relationship between the Lagrangian function L and physical quantities, namely, $L = T - V$, where T is the kinetic energy of the particle and V is its physical potential energy. But when potential energy depends on particle velocity as well as position the formalism loses direct contact with physics and requires a Lagrangian function employing a so-called "generalized potential energy" having nothing to do with, and quite distinct from, the

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