

**Bernard Feder**

**THE  
COMPLETE  
GUIDE TO  
TAKING  
TESTS**



**How to earn higher scores  
on any exam – with  
test-taking strategies  
that really work**

# The Complete Guide To Taking Tests

**Timothy B. Rogers**



## **The Complete Guide To Taking Tests:**

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web an answer key document has been included and is available for download some additional information to be aware of for teachers demonstrations 1 and 5 will not produce a chemical reaction five of the reactions will produce indicators of a chemical reaction for students to observe

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web identifying chemical reaction types key part 1 identify the type of reaction 1  $\text{Na}_3\text{PO}_4 + 3\text{KOH} \rightarrow 3\text{NaOH} + \text{K}_3\text{PO}_4$  double displacement 2  $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2\text{LiCl}$  double displacement 3  $\text{C}_6\text{H}_{12} + 9\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$  combustion 4  $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$  single displacement 5  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$  decomposition 6  $\text{P}_4 + 3\text{O}_2 \rightarrow 2\text{P}_2\text{O}_3$

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web chemical equations can also be used to represent physical processes write a chemical reaction for the boiling of water including the proper phase labels chemical equations can also be used to represent physical processes write a chemical reaction for the freezing of water including the proper phase labels explain why  $4\text{Na} + \text{S} \rightarrow 2\text{Na}_2\text{S}$

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web name score date balancing and identifying chemical reactions a identify each type of chemical reactions 1  $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$  2  $\text{Sr} + \text{NO}_3 \rightarrow \text{Sr} + \text{NO}_2 + \text{O}_2$  3  $2\text{LiBr} + \text{F}_2 \rightarrow 2\text{LiF} + \text{Br}_2$  4  $2\text{Na} + \text{F}_2 \rightarrow 2\text{NaF}$  5  $\text{F}_2 + \text{Br}_2 \rightarrow 2\text{FBr}$  b balance the chemical reactions and identify each type

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web 1  $3\text{NaBr} + 1\text{H}_3\text{PO}_4 + 1\text{Na}_3\text{PO}_4 + 3\text{HBr}$  type of reaction double displacement 2  $3\text{Ca} + \text{OH}_2 + 1\text{Al}_2\text{SO}_4 + 3\text{CaSO}_4 + 2\text{Al} + \text{OH}_3$  type of reaction double displacement 3  $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$  type of reaction single displacement 4  $1\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$  type of reaction combustion 5  $2\text{PbSO}_4 + 2\text{PbSO}_3 + 1\text{O}_2$  type of reaction

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web a chemical reaction is a process in which one or more substances are converted to one or more different substances the starting substances are called the reactants and the new substances that form are called the products

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web study guide chemical reactions 1 give an example of a chemical reaction leaves turning color fireworks food spoiling 2 differentiate between a physical and chemical change physical does not change the identity of the matter can be changed back easily chemical new substance is created with new physical and chemical