



The Bastards Tale

RJ Shavelson



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web log $[\text{h}^+] = 10^{-4.05} = 8.91 \times 10^{-5} \text{ mol dm}^{-3}$ log pka ph 3.75 4.05 0.30 10.0 30.0 50.0 and hcooh 0.1 so hcoo^- 0.200 mol dm⁻³ amount vol x

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