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# Tutorial Roblox Adventure Game

**SB Merriam**



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web on the basis of the entrance of the last electron either into 4 f orbital or 5 f orbital these elements are classified into two categories called lanthanides and actinides respectively lanthanides the name lanthanides comes from lanthanum la z 57 because these elements are closely related to lanthanum

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web the actinides are elements 89 to 103 and fill their 5f sublevel progressively actinides are typical metals and have properties of both the d block and the f block elements but they are also radioactive lanthanides have different chemistry from transition metals because their 4f orbitals are shielded from the atom's environment

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make up the f block of the periodic table the lanthanides are the elements produced as the 4f sublevel is filled with electrons and the actinides are formed while filling the 5f sublevel generally speaking the lanthanides

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