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# Time To Stand

**CH Cherryholmes**



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 $\text{NH}_3$  in liters needed at  $20^\circ\text{C}$  and 25 atm to react with 150 kg of  $\text{H}_2\text{SO}_4$

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12  $\text{CO}_2$  using the following equation to answer questions 8 11 2  $\text{C}_6\text{H}_{10} + 17 \text{O}_2 \rightarrow 12 \text{CO}_2 + 10 \text{H}_2\text{O}$  key solutions for the  
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