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# The War Between the Classes

# War Between The Classes

**Marcel A. Müller**



## **War Between The Classes:**

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web balancing equations the chemical equation described in section 4 1 is balanced meaning that equal numbers of atoms for each element involved in the reaction are represented on the reactant and product sides this is a requirement the equation

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web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide  $\text{CaO}$  and  $\text{H}_2\text{O}$  to form aqueous calcium hydroxide b is it possible to balance the equation if you incorrectly identify the product as  $\text{CaOH}$  and if so what is the equation

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water and carbon dioxide 2 write down the number of atoms per element

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