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# Soviet Political Elites

**SA Dillow**



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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions

regardless of the source of these ions these molecular and  
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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh<sub>4</sub> aq cr<sub>2</sub>o<sub>7</sub><sup>2-</sup> aq ag<sub>2</sub>cr<sub>2</sub>o<sub>7</sub> s 2nh<sub>4</sub> aq 2f aq because two nh<sub>4</sub> aq

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