



# Pro Future Of Work

**RC Schank**



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$$m = \frac{M}{1000d}$$

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web 1. molarity  $M = \frac{m}{V}$  mass molar mass  $x = \frac{4500}{1.0825} = 4149579 \text{ g/mol}$   $x = \frac{0.0129}{2} = 0.00645 \text{ mol}$  molality  $m = \frac{0.825}{1.419579} = 0.581158 \text{ mol/kg}$  0.00581158 mol/kg 4500 kg 0.0129 m 3 mole fraction  $n_2 = \frac{4.0825}{4.0825 + 141.9579} = 0.0281158$  mol % 4500 g 18.015 g/mol 24.97918401 mol

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m mix m 1 v 1 m 2 v 2 v 1 v 2 1 5 x 480 1 2 x 520 480 520 720 624 1000 1 344 m q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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