



# Soul Consciousness

**Yicheng Fang**



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web polyatomic ions are charged groups of atoms an example is ammonium ion nh<sub>4</sub> it has five atoms one nitrogen and four hydrogens that share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using subscripts for example ammonium chloride is nh<sub>4</sub>cl and ammonium sulfide is nh<sub>4</sub> 2s

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web rules for naming ionic compounds containing polyatomic ions polyatomic ions are ions which consist of more than one atom for example nitrate ion no<sub>3</sub> contains one nitrogen atom and three oxygen atoms the atoms in a polyatomic ion are

usually covalently bonded to one another and therefore stay together as a single charged unit

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web the formulae of compounds containing polyatomic ions are worked out in a similar way to single atom ions except when there is more than one polyatomic ion then its formula is written