

- ✔ Treat your **HOTEL ROOM** like it is your own **HOME**
- ✔ Know the **LOCAL CUSTOMS** and follow them
- ✔ Find out the real **MONETARY VALUE** of things and adhere to those prices
- ✔ Ask for permission before taking somebody's **PHOTO**
- ✔ Respect local **TRADITIONS** and **RELIGION**
- ✔ Learn some basic words and phrases in the local **LANGUAGE**
- ✔ Support the local shops and **ARTISANS**
- ✔ Report **CHILD** sex tourism

- ❌ Don't throw away or leave behind any **EXPENDABLES** that you brought
- ❌ Don't buy illegal **WILDLIFE** products and **HERITAGE** antiques
- ❌ Don't encourage **BEGGING** by giving money to paupers and **STREET KIDS**, rather donate to a local **NGO**
- ❌ Don't overly display your "first-world" **GADGETS**
- ❌ Don't be condescending or arrogant when **TALKING** to locals
- ❌ Don't **JUDGE** an entire culture based on a single individual
- ❌ Don't agree to **BRIBE** officials
- ❌ Don't **LITTER**, even if locals do



Offset your **CARBON FOOTPRINT**



Minimize air travel in favor of land or sea **TRANSPORT**



Join a local **NGO** or do some **VOLUNTARY** work
 ↳ learn how you can make a difference



Set a **GOOD EXAMPLE** for locals and fellow travelers
 ↳ how you behave will influence how locals see future travelers



Curb your **FREEDOM** of **SPEECH** about government and religion
 ↳ not doing so could get you and your local friends in serious trouble



Stay in family-run hotels/hostels or do a **HOMESTAY**



Carry **REUSABLE CONTAINERS**
 ↳ use them at street food and drink stalls instead of the disposable ones provided



Sustainable Travel Tutorial

Yan Bai



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Embark on a breathtaking journey through nature and adventure with Crafted by is mesmerizing ebook, Witness the Wonders in **Sustainable Travel Tutorial** . This immersive experience, available for download in a PDF format (*), transports you to the heart of natural marvels and thrilling escapades. Download now and let the adventure begin!

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jan 3 2023 a net ionic equation is a chemical equation that shows only the species participating in a chemical reaction and omits the spectator ions spectator ions are ions that

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations

requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction

therefore the net ionic equation is $2\text{Cl}^- + 2\text{Ag}^+$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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may 11 2017 in the net ionic equation any ions that do not participate in the reaction called spectator ions are excluded as a result the net ionic equation shows only the species that

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$1\text{K}_3\text{PO}_4 + \text{Al}(\text{NO}_3)_3 \rightarrow 3\text{KNO}_3 + \text{AlPO}_4$ s 3 K^+ aq 3 NO_3^- aq 3 Cl^- aq Al^{3+} aq 3Na^+ aq AlPO_4 s 3 K^+ aq net ionic Al^{3+} aq PO_4^{3-} aq 3 Mg^{2+} aq 3 $\text{Mg}(\text{OH})_2$ s

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oct 2 2019 the net ionic equation is a chemical equation for a reaction that lists only those species participating in the reaction the net ionic equation is commonly used in acid base

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aug 10 2023 write the full ionic and net ionic equations for this reaction solution the elemental metals magnesium on the reactant side copper on the product side are neutral

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds cl aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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