



# The Underneath

**Jiyuan Zhang**



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b we will use one graduated cylinder to represent the amount of the reactant a and a second graduated

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hscn present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species  
 $\text{mol Fe} = 3.170 \times 10^{-5} \text{ mol}$   
 $\text{mol Fe} = 3.850 \times 10^{-4} \text{ mol}$

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