

DAILY SCALE TRENDS FOR FAT LOSS

STAY *fit* MOM

WHEN YOU WANT TO GIVE UP

WHEN YOU'RE FEELING MOTIVATED



**BODYWEIGHT WILL FLUCTUATE DAILY.
LOOK FOR A DOWNWARD TREND OVER TIME.**

Weight Loss Trending

Mark Steyvers



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web eastern lowland gorilla gorilla gorilla graueri is slightly larger in size weighing up to 220 kg 484 lbs and darker in coloration than the western lowland gorilla they live in the rainforests of central africa mountain gorilla gorilla gorilla beringei is the largest and rarest of all three subspecies adult males may weigh over 227 kg

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web the males grow to around 5 ½ feet tall and weigh around 400 pounds the females grow to 4 ½ feet tall and weigh around 200 pounds gorillas have long arms even longer than their legs they use their long arms to knuckle walk this is where they use the knuckles on their hands to walk on all fours

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web the gorilla is the largest of all primates western lowland adult male gorillas gorilla gorilla gorilla are about 1 7 m 66 9 in in body length and weigh about 169 5 kg 374 lbs adult females are about 1 5 m 59 1 in in body length and weigh about 71 5 kg 158 lbs

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web gorillas are mainly herbivores and their diet mostly consists of bamboo fruit and leafy plants though western lowland gorillas also eat small insects adult gorillas can eat up to 30kg of food each day as roaming herbivores gorillas play a vital role in seed dispersal

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robust with males weighing around 168 kilograms 370 lb the hair is significantly lighter in color than that of the eastern
gorilla gorilla beringei and geographically isolated from them in a region at the

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officials to death for crimes committed during the nazi regime each of the four allied nations the united states great britain
the soviet union and france supplied a judge and a prosecution team to the imt for the trial of

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representatives of the defeated nazi germany for plotting and carrying out invasions of other countries and atrocities against
their citizens in

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carried out in nuremberg germany between 1945 and 1949

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web unable to find a suitable venue in the destroyed nazi capital the court soon moved to the city of nuremberg nürnberg in bavaria where the highest profile cases were heard in the aptly named palace of justice between november 20 1945 and august 31 1946

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web aug 8 2023 nurnberg trials a series of trials held in nurnberg germany in 1945 46 in which former nazi leaders were indicted and tried as war criminals by the international military tribunal the court rejected the defense that only countries could perpetrate war crimes and found most of the original 24 defendants guilty

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web at 10 00 am on 20 november 1945 sir geoffrey lawrence the presiding judge at the first of the nuremberg trials opened proceedings at what he described as a trial that was unique in the

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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boiling point determination lab report conclusion objectivesto correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objekte can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound's normal boiling point refers to its boiling point at a pressure of 760 mm Hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment I aim to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a Bunsen burner boiling

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there are a variety of methods by which a sample's boiling point can be determined including distillation reflux and by using a Thiele tube the most straightforward method uses a Thiele tube and has the advantage of using less than 0.5 mL of material